

Role of Nitric Oxide in Cancer Biology

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The role of nitric oxide (NO) in tumorigenesis is multifactorial. NO can participate in the complicated process of carcinogenesis by mediating DNA damage in early phases of tumorigenesis, as well as support tumor progression through the induction of angiogenesis and suppression of the immune response. This paper addresses the effects of NO on transcriptional regulation following DNA damage and cyclooxygenase expression in the multistep process of tumorigenesis.

Keywords: Nitric oxide, tumor, p53, cyclooxygenase

INTRODUCTION

A small, chemically simple and highly toxic gas, nitric oxide (NO) seems an unlikely biological jack-of-all-trades, as most of the body's functions are indeed regulated by extraordinarily large and complex proteins and compounds. NO exists in the gaseous state under normal atmospheric conditions. Over the past decade, it has been realized that this diatomic radical plays a variety of regulatory functions *in vivo*.^[1] Interest in the

importance of NO in medicine and biology escalated after the first paper on this topic, published in 1985 by Stuehr and Marletta, demonstrated NO synthesis by a mammalian cell.^[2] Unsurprisingly, NO was named "Molecule of the Year" in 1992 in recognition of the tremendous significance of its role in biological systems.^[3]

Nearly every cell type studied thus far has the capacity to synthesize NO by one of three distinct nitric oxide synthase (NOS) isoforms.^[1] The effects brought forth by NO can vary greatly, depending on the site where NO is synthesized, the amount of NO formed, and the targets within the local environment. Beneficial responses, such as vasodilatation, inhibition of platelet aggregation, relaxation of smooth muscles and signal transmission in neuronal cells are induced by the production of small quantities of NO by constitutive NOS, namely endothelial and neuronal isoforms. These constitutive isoforms, which are expressed continuously, generate NO in conditions where intracellular Ca²⁺ level is increased and calmodulin is activated.^[1] In contrast, the action of

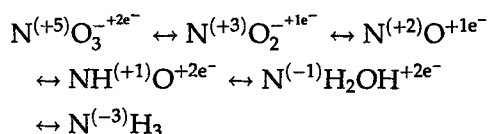
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inducible NOS is independent of intracellular Ca^{2+} level and leads to the synthesis of high local concentrations of NO for prolonged periods of time. This isoform of NOS produces much larger quantities of NO in response to cytokines and endotoxins, which have been shown to be cytostatic or cytotoxic to tumor cells and a range of microorganisms.^[4] The effects of NO are not all beneficial, as this free radical has also been identified as a deleterious agent in numerous pathophysiological conditions, including cancer. The anti-tumor properties of NO are highlighted by some authors, while others implicate NO in tumor promotion.

The objective of this review is to discuss these seemingly contradictory roles of NO in early and late stages of human carcinogenesis.

BIOCHEMICAL INTERACTIONS OF NO

The redox properties of NO influence its reactivity. The redox relationship of NO with other nitrogen oxides indicates that NO occupies a central and unique position in the redox scheme.^[5]



Being a free radical, NO is capable of reacting with other free radicals. In basic conditions, NO is capable of oxidizing sulfhydryl groups, reacting with amines in organic solvents,^[6] and with metal-containing proteins.^[7,8] It is crucial to understand the NO chemistry, before one can fully grasp its physiological functions as well as its toxicity. The role of NO as a physiological messenger molecule and as a cytotoxic effector molecule under appropriate conditions is dependent on its chemical properties as well as concentration. Reactive nitrogen oxide species (RNOS) are produced by the reaction of NO with either molecular oxygen, one of the most common yet

biologically reactive molecules, or with superoxide. Numerous biological molecules have been proven to be oxidized by RNOS,^[9] therefore possibly accounting for certain types of NO-mediated toxicity.

The reactivity of NO is one of the primary determinants of its final effect on a given biological system. NO acts both directly and indirectly.^[10] NO acts directly by reacting chemically with a specific biological target, mostly in conditions where constitutive NOS isoforms generate low levels of NO.^[10] Thus, low levels of NO can react directly with certain heme-containing proteins such as guanylate cyclase, oxyhemoglobin, and cytochrome P450.^[11] On the other hand, the reactive nitrogen oxide species mediate certain chemical reactions, where the local concentration of NO, produced by iNOS alone, is high. These constitute the indirect effects of NO. These chemical reactions bring about nitrosative and oxidative stresses in biological systems, both of which yield different types of DNA mutations.^[11]

Nitrosamines generated in biological systems during nitrosative stress are potentially carcinogenic. It has been shown that these carcinogenic nitrosamines can be generated in the acidic conditions of the stomach.^[12-17] Although NO itself does not directly interact with DNA or proteins, high concentrations of NO generated during inflammation as a response to any pathogen, can lead to the formation of RNOS, such as N_2O_3 .^[18] During inflammation, stimulated macrophages and neutrophils express iNOS and generate large amounts of NO. This promotes the formation of RNOS, which can then nitrosate amines.^[19,20] Hence such nitrosation reactions appear to occur *in vivo*. Nitrosamines, formed under such conditions of chronic inflammation, can in turn lead to cancer.^[21,22] Cells exposed to NO exhibit lesions consistent with the chemistry of deamination. This involves the conversion of cytosine to uracil, guanine to xanthine, methylcytosine to thymine, and adenine to hypoxanthine.^[23,24] Single-stranded DNA appears to be far more susceptible to nitrosation than double-stranded DNA,^[25]

suggesting that deamination occurs mainly during the replication and transcription of DNA. Such mechanisms involving the nitrosation of nucleic acids may contribute to *in vivo* spontaneous deamination.

Oxidative stress induced by RNOS is thought to be mediated primarily by the formation of peroxynitrite,^[11,20,26] since this can induce DNA strand breaks *in vitro*.^[27]

NO and RNOS may also act indirectly by affecting the enzymatic activity of several DNA repair proteins. RNOS have a high affinity for amino acids containing thiol residues in DNA repair proteins and they can induce nitrosation in their active sites.^[28–20] Another important reaction is the inhibition of DNA-binding proteins containing zinc finger motifs^[31] in the presence of NO under aerobic conditions. This occurs perhaps through the nitrosation of thiols by N₂O₃, subsequent ejection of zinc, and the consequent degradation of the structural integrity of the protein.^[32]

Exposure of cells to NO results in an increased number of single-strand DNA breaks.^[24] It has been shown that the effects of exposure to NO or RNOS include the inhibition of DNA ligase activity, leading to an accumulation of DNA breaks that occur either during transcription or repair. This increase in DNA breaks caused by the NO-mediated inhibition of ligase could in turn activate the tumor suppressor gene, p53.^[33]

ROLE OF NO IN THE PROCESS OF CARCINOGENESIS

The etiology of cancer involves many factors, with the earliest oncogenic alterations occurring possibly decades before transformation. Tumors are formed in a process that consists of several stages: initiation, promotion, and progression. Via both its direct and indirect paths of action, NO is implicated in tumor promotion as well as tumor suppression. Evidently, NO plays a

complex and sometimes contradictory role in carcinogenesis. The association between increased NOS expression and tumor progression as well as metastasis is proposed by some studies,^[34–41] while others draw a link between tumor invasiveness and aggressiveness and decreased NOS activity.^[42]

The involvement of NO in tumor biology has been a topic of much study in recent years. Interactions between the endothelial cells of tumor vasculature, tumor-infiltrating immune cells such as T lymphocytes and macrophages, and the tumor cells themselves play a regulatory role in tumor growth. The production of NO by most of these cellular components has been demonstrated.^[43–46]

The role of NO in cancer is two-faceted: when produced at high concentrations, it has cytotoxic and cytostatic properties, resulting in DNA damage, the initial step in carcinogenesis. Tumor growth is enhanced via NO's anti-apoptotic effects and its role in the creation of neovasculation, under conditions of low to moderate NO concentration.^[47]

INITIATION PHASE OF TUMORIGENESIS

The stage in tumor growth, the location of NO production and the concentration of NO determine the effects of NO in tumor biology. Tumor growth is a multistep process including the initiation phase and the enlargement of an established tumor. Accumulation of mutations occurs in tissues exposed to prolonged high NO concentration, as a result of NO itself, or through the facilitation of other genotoxic agents, for instance during episodes of chronic inflammation or through exposure to mutagens in the environment. Inhibition of DNA synthesis via the inhibition of ribonucleotide reductase activity^[48–50] and inhibition of mitochondrial respiration by reacting with iron-sulphur proteins such as acconitase^[51–54] are a few examples of the numerous

mechanisms of NO cytostasis. By the formation of peroxyxynitrite, carcinogenic nitrosamines and the inhibition of systems required for the repairing of DNA damage, NO may aggravate DNA damage.^[28,29,32,55-58] Exposure of cells to NO results in an increased number of single-strand DNA breaks.^[24] Furthermore, NO or RNOS induced DNA damage, activates the tumor suppressor gene, p53.^[33]

In the initiation phase of tumorigenesis, initial DNA damage causes the accumulation of the tumor suppressor gene, p53.^[33] This acts as a checkpoint control in the cell cycle, permitting the repair of damaged DNA.^[59] As p53 activation causes a block in G1/S transition in the cell cycle, apoptosis results in the case of severe DNA damage.^[60,61] p53 thus plays an active role in the cellular response to endogenously produced DNA damage due to NO and RNOS.

It can be seen that NO-induced DNA damage results in wild-type p53 accumulation. Furthermore, it has been reported that increased expression of wild-type p53 downregulates iNOS expression via a negative feedback loop that results in the reduced production of NO.^[59] In addition, prolonged transcription of wild-type p53 may cause methylation of p53 tumor suppressor gene, which favors G:C → A:T transitions in this gene due to NO.^[21,23,24] It has been shown that G:C → A:T transition is one of the most common mutations in the p53 tumor suppressor gene in early phases of human carcinogenesis.^[62] As a result of this process, the mutated p53 gene cannot protect cells and tissues from mutation in the event of severe DNA damage. Therefore, cells and tissues are allowed to multiply in the event of severe DNA damage since the mutated p53 gene loses its principal function in cell cycle control.

As the actions of NO and the p53 tumor suppressor gene in cell cycle control are inhibited, a functional derangement of the cell cycle and uncontrolled cellular proliferation ultimately result, thus contributing to the initial stage of tumorigenesis.

TUMOR-PROMOTING PROPERTIES OF NO IN ADVANCED TUMORS

When the tumor cells have reached a certain number, hypoxia occurs in the enlarging tumor and the creation of neovasculature becomes necessary for further tumor growth and survival. Wild-type p53 is a known inhibitor of tumor angiogenesis,^[63] but the presence of hypoxia, arising from enlargement of the tumor, has been described as a selecting factor for mutant p53. Mutational inactivation of p53 generates a cell population that can tolerate the genotoxicity and cytostasis of sustained NO production.^[64] Furthermore, low to moderate NO concentrations may even have anti-apoptotic properties in these cells. Therefore, the loss of p53 function in p53-mutated cells would permit both the growth of tumor cells as well as the release of angiogenic factors such as vascular endothelial growth factor in the presence of moderate NO concentrations.^[65] At this stage, the production of iNOS, induced by hypoxia, stimulates the production of NO by tumor cells.^[66] In this respect, NO production may be a part of the angiogenic switch in developing tumors, without which a lack of vascularization would result, hence limiting the size of the developing tumors (Figure 1). Thus, hypoxia and cytokines (e.g. tumor necrosis factor) produced by tumor cells induce iNOS expression in these cells. Subsequent production of NO then promotes tumor growth by stimulating angiogenesis,^[47,67,68] increasing vascular permeability^[69-72] and suppressing the immune response.^[73,74] The suppression of leukocyte proliferation and infiltration is another systemic effect of NO of relevance in cancer biology. Several reports have indicated that NO produced by tumor cells may prevent the infiltration of leukocytes^[73] and leukocyte adhesion.^[68] Thus, NO, in addition to increasing the vascular permeability, downregulates the expression of some adhesion molecules, which are important for inflammatory and immune cell adhesion to vascular endothelium.^[69]

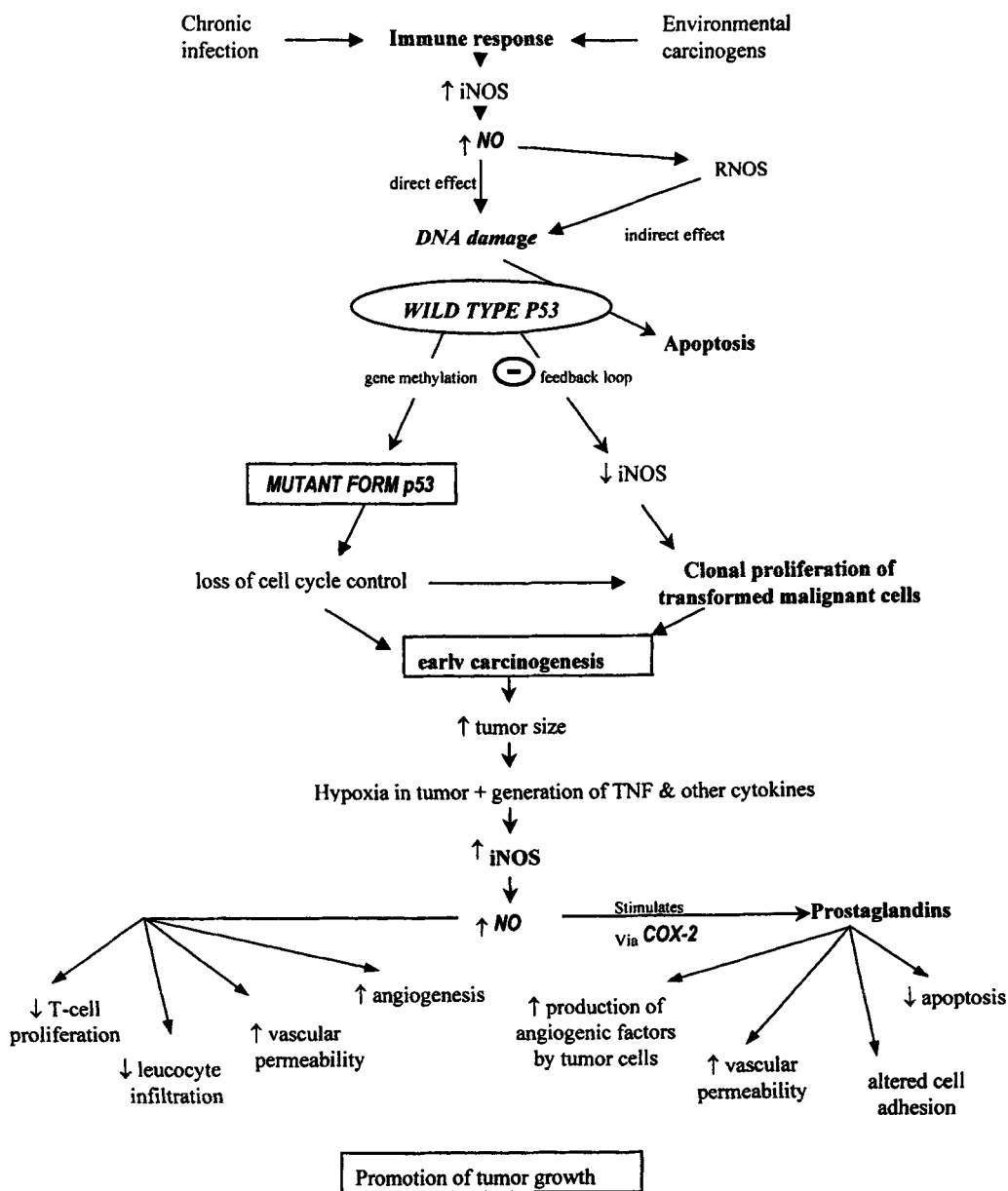


FIGURE 1 The involvement of molecular parameters in initial and late stages of carcinogenesis.

Another mechanism by which NO may promote tumor growth is by modulating the production of prostaglandins.^[75] NO activates COX-2,^[76] which in turn, stimulates the production of proangiogenic factors and prostaglandins. Prostaglandins increase vascular permeability supporting the development of neovasculature in

tumors.^[77,78] It has also been shown that the overexpression of COX-2 alters cell adhesion and protects cells against apoptosis by increasing the Bcl-2 protein production.^[79,80] All these effects of NO and prostaglandins generated by COX-2 facilitate further tumor growth.

SUMMARY

Normal tissue homeostasis is maintained through the regulation of cell proliferation and apoptosis. It has been shown that in normal cells, wild-type p53 suppress both basal and cytokine-induced iNOS via a negative "feedback loop".^[59] However, in tumor cells which express mutant p53, iNOS expression would be unchecked. These cells, lacking a functional p53 protein, are prevented from being eliminated in the event of severe DNA damage, and therefore proliferate more aggressively. Hypoxia, arising from enlargement in tumor size, has been described to be a selective factor for mutant p53 as hypoxia has both genotoxic as well as angiogenic properties.

The loss of wild-type p53 function or the expression of mutant p53 in the tumor cells would permit both the growth of tumor cells in the presence of moderate NO concentrations as well as the release of angiogenic factors such as the vascular endothelial growth factor. Besides supporting the development of neovasculature, increased iNOS levels in several solid tumors, such as human breast, brain, head and neck, and colon cancers,^[41,65,81,82] may lead to a wild-type p53-mediated growth arrest in the epithelial cells close to the source of NO production. The resulting growth inhibition would subsequently exert a strong selection pressure, allowing those cells that contain mutant p53 to proliferate faster than those containing wild-type p53.^[65] It has been reported that mutant p53-positive tumors are associated with poor prognosis and resistance to chemotherapy.^[83-86] Furthermore, clonal selection and growth of these mutant p53-containing cells are further supported by the combination of prostaglandins and NO-induced angiogenesis, increased vascular permeability, immune suppression and reduced apoptosis (Figure 1).

It has been proposed that tumor-associated NO production, as well as cyclooxygenase-2 overexpression, promotes cancer progression by providing a selective growth advantage to tumor cells with mutant p53.^[65] This is consistent with

the hypothesis that NO and COX-2 are the cancer-promoting factors in human carcinogenesis. Therefore, according to this hypothesis, it may be possible in the future that the inhibitors of iNOS and COX-2 may have a therapeutic effect on these human tumors, which could be a direction of future research in the area of cancer treatment.

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